Pharmacokinetics Metabolism

## Learning object

- Know the processes involved in ADME of drugs
- Know how these processes may affect the action of xenobiotics
- Appreciate how these processes can affect the outcome of the treatment of patients with drugs
- Appreciate how differences in these processes between patients can affect therapy
- Know how these processes have been exploited to improve therapy
- Be able to exemplify the above

## BIOTRANSFORMATION (Metabolism)

• Chemical alteration of the drug in the body is called as biotransformation.

It is needed to render
 Nonpolar (lipid-soluble) compounds
 polar (Water soluble)
 excreted in urine.

Most Polar (hydrophilic drugs) e.g.
 Streptomycin

little biotransformed excreted unchanged. • The primary site for drug metabolism is liver

- Others sites are:-
  - kidney
  - intestine
  - lungs
  - Plasma



#### 2. Active drug — Active metabolite

#### Eg.

 Morphine — Morphine-G glucuronide (Active) (active) 3. Inactive drug Active drug

• eg. Prodrug such as:

Levodopa → Dopamine (Inactive) (Active)

- Advantages of prodrug over the active form
  - more stability,
  - better bioavailability
  - activated selectively at the site of action
  - less side effects and toxicity

#### **Classification of biotransformation reaction**

I. Nonsynthetic/Phase I/Functionalization reactions

active or inactive - metabolite

- 5 types
  - Oxidation
  - Reduction
  - Hydrolysis
  - Cyclization
  - Decyclization



#### • VII Types:

- (i) Glucuronide conjugation
- (ii) Acetylation
- (iii) Methylation
- (iv) Sulfate conjugation
- (v) Glycine conjugation
- (vi) Glutathione conjugation
- (vii) Ribonucleoside/nucleotide synthesis

## **Nonsynthetic reactions**

(i) Oxidation: This reaction involves addition of oxygen/negatively charged radical or removal of hydrogen/positively charged radical.

Eg. Barbiturates, phenothiazines, imipramine, ibuprofen, paracetamol

#### (ii) Reduction: This reaction is the converse of oxidation

Involves cytochrome P-450 enzymes working in the opposite direction.

• Eg. Chloralhydrate, chloramphenicol, halothane, warfarin.

## (iii) **Hydrolysis** This is cleavage of drug molecule by taking up a molecule of water.

Ester + H<sub>2</sub>O 
$$\xrightarrow{esterase}$$
 Acid + Alcohol

Hydrolysis occurs in liver, intestines, plasma and other tissues.

• Eg: Procaine, lidocaine, procainamide, aspirin.



(v) **Decyclization**: This is **opening up of ring structure** of the cyclic drug molecule e.g. Barbiturates, phenytoin.

![](_page_14_Figure_1.jpeg)

## **Phase II: Synthetic reactions**

 These involve conjugation of the drug or its phase I metabolite with an endogenous substrate to form a polar highly ionized organic acid which is easily excreted in urine or bile.

## **1. Glucuronide conjugation**

 This is the most important synthetic reaction carried out by a group of UDP-glucuronosyl transferases (UGTs).

 Compounds with a hydroxyl or carboxylic acid group are easily conjugated with glucuronic acid which is derived from glucose.

• Eg. chloramphenicol, aspirin, paracetamol, morphine, metronidazole.

![](_page_17_Figure_0.jpeg)

## (ii) Acetylation

 Compounds having amino or hydrazine residues are conjugated with the help of acetyl coenzyme-A e.g. sulfonamides, isoniazid.

It shows genetic polymorphism (slow and fast acetylators).

## (iii) Methylation

- The amines and phenols group of drugs can be methylated
- Methionine and cysteine acting as methyl donors e.g. histamine.

![](_page_19_Figure_3.jpeg)

## (iv) Sulfate conjugation

• The phenolic compounds and steroids are sulfated by sulfotransferases (SULTs)

• e.g. chloramphenicol, sex steroids

## (v) Glycine conjugation

 Salicylates and other drugs having carboxylic acid group are conjugated with glycine,

• It is not a major pathway of metabolism.

## (vi) Glutathione conjugation

 Highly reactive quinone or epoxide intermediates conjugate with glutathione
 mercapturate.

 It serves to inactivate highly reactive quinone or epoxide intermediates formed during metabolism of certain drugs e.g. paracetamol

![](_page_23_Figure_0.jpeg)

![](_page_24_Figure_0.jpeg)

#### (vii) Ribonucleoside/nucleotide synthesis

• This pathway is important for the activation of many purine and pyrimidine antimetabolites used in cancer chemotherapy.

#### Simultaneous or sequiential metabolism of drugs

![](_page_26_Picture_1.jpeg)

## Drug Metabolizing Enzymes

- Microsomal enzymes
- Non-microsomal enzymes

### **Microsomal enzymes**

 Location: smooth endoplasmic reticulum, primarily in liver, also in kidney, intestinal mucosa and lungs.

- Eg. of ME
  - Monooxygenases,
  - cytochrome P 450,
  - glucuronyl transferase etc.

- ME catalyse
  - oxidations,
  - reductions,
  - hydrolysis
  - and glucuronide conjugation
- Microsomal enzymes are inducible by
  - drugs
  - diet.

## Non microsomal enzymes

• Location: Cytoplasm, mitochondria of Liver & Plasma.

#### Eg of NME

- Flavoprotein oxidases,
- esterases,
- amidase
- And conjugases.

- Reaction catalysed are:
  - Some oxidations and reductions,
  - many hydrolytic reactions
  - and all conjugations excep: glucuronidation.

- The nonmicrosomal enzymes are
  - not inducible
  - but many show genetic polymorphism.

![](_page_32_Figure_0.jpeg)

more susceptible to many drugs e.g. chloramphenicol, opioids.

Low dose of drug used usual dose prod toxicity

#### **INHIBITION OF DRUG METABOLISM**

- 2 types
  - 1. Competetive enzyme inhibition
  - 2. Non competative enzyme inhibition
- One drug can competitively inhibit metabolism of another if it utilizes the same enzyme or cofactors.

• Occurs in a dose related manner and can precipitate toxicity of the object drug.

Metabolism of drugs with high hepatic extractionis
 dependent on liver blood flow.

eg. Propranolol decreases hepatic blood flow.

 Reducese rate of lidocaine metabolism

#### **MICROSOMAL ENZYME INDUCTION**

Many drugs, insecticides and carcinogens interact with DNA

Increase the synthesis of microsomal enzyme protein (cytochrome P-450 and glucuronyl transferase).

**Increases** rate of metabolism.

 Eg. Phenobarbiton - inducers of Cyp3A and Cyp2D6

Increase the rate of metabolism by 2-4 fold.

- Polycyclic hydrocarbons like 3-methyl cholanthrene and benzopyrene found in
  - cigarette smoke
  - charcoal boiled meat
  - and industrial pollutants

![](_page_37_Picture_4.jpeg)

![](_page_37_Picture_5.jpeg)

Induce CYP1A isoenzymes.

- Other important enzyme inducers are:
  - Chloral hydrate,
  - phenylbutazone,
  - griseofulvin.

#### **Consequences of microsomal enzyme induction**

1. <u>Decrease intensity or duration of action</u> of drugs that are inativated by metabolism.

Eg. Oral contraceptives

2. <u>Increase intensity of action of drugs</u> that are activeted by metabolism.

Eg. Acute paracetamole toxicity

**3.** <u>Tolerance</u>: In case of autoinduction.

Eg. Carbamazapine.

4. <u>Precipitation of acute intermittent porphyria-</u> disease in which an important part of hemoglobin, called heme, is not made properly.

5. Interfere with adjustment of dose of another drug.

6. Interfere with chronic toxicity testing in animal.

7. It affect metabolism of other drugs eg.

1. Phenytoin

2. Warfarine

3. oral contraceptive

4. chloramphenicol etc.

## Summary

- Decrease the intensity
- Increases intensity of drug
- Tolerance
- Prophyria
- Interfere with adjustment of dose
- Interfere with chronic toxicity testing

#### FIRST PASS (PRESYSTEMIC) METABOLISM

#### Definition:

Metabolism of drug during its passage from site of absorption into systemic circulation.

1. Orally administered drug

![](_page_43_Picture_4.jpeg)

Come in contact with intestinal & liver enzyme

# 2. It is also occure in skin & lungs

Low	Intermediate	High	
		not given orally	high oral dose
Phenobarbitone	Aspirin	Isoprenaline	Propranolol
Phenylbutazone	Quinidine	Lidocaine	Alprenolol
Tolbutamide	Desipramine	Hydrocortisone	Verapamil
Theophylline	Nortriptyline	Testosterone	Salbutamol
Pindolol	Chlorpromazine		Glyceryl trinitrate
Isosorbide	Pentazocine		Morphine
mononitrate	Metoprolol		Pethidine

#### Table 3.1: Extent of first pass metabolism of some important drugs

Attributes of drugs with high first pass metabolism:

(a) **Higher oral dose** than sublingual or parenteral dose.

(b) Marked individual variation in the oral dose due to differences in the extent of first pass metabolism.

## (c) Oral bioavailability is apparently increased in patients with severe liver disease.

(d) Oral bioavailability of a drug is increased if another drug competing with it for first pass metabolism (e.g. Chlorpromazine and propranolol).

![](_page_48_Figure_0.jpeg)

## **Excretion of drug**

- Drug can be excreted in their unchanged form or as metabolites by
  - Kidney
  - lungs
  - intestine
- And to lesser extend by the
  - sweat
  - salivary
  - and mammary glands.

• More water soluble a drug or meatabolite

More efficiently it is elimineted from the body.

<u>Renal excretion:</u>

3 major process involve in the excretion of drugs by the kidney

- 1. Glomerular filtration
- 2. Tubular secreation
- 3. Tubular reabsorption

Sylvia S. Mader, Inquiry into Life, 8th edition. Copyright © 1997 The McGraw-Hill Companies, Inc. All rights reserved.

![](_page_51_Figure_1.jpeg)

Glomerular filtration & Tubular secreation

Remove the drug from plasma

Tubular reabsorption

• The net excretion of drug depends therfore on sum total of three processes.

## **Glomerular filtration**

 By this process drug molecules diffuse out of the blood --- <u>filtrating</u> the glomeruli of the nephron -passes into the tubuls of the kidney.

Drug with molecular wt of 500 or less

Unimpeded diffusion of the drug

![](_page_54_Figure_0.jpeg)

#### Result:

• All the drug which are **not protein bound** appear in the glomerular filtrate.

- Glomerular filtrate 120 L / day
- But daily urine volume 1-2 % of this volume.
- Factors influence renal excretion by glomerular <u>filatration</u>:

![](_page_55_Figure_3.jpeg)

### **Tubular secretion**

- Acidic drugs eg. salicylates & penicilline.
- Basic drugs eg. Adrenaline & noradrenaline.

 Many drugs which are secreted by tubuls are also filtered by the glomeruli

Having very short duration of action

• Eg Penicilline

## How to overcome short duration of action of penicillin due to tubular secreation:

• Since tubular secretion required energy, it can be inhibited by certain drug.

• Eg. Probenicide is often combined with penicillin

inhibit active secretion process of penicilline.

Prolonging the duration of action of penicilline

## **Tubular reabsorption**

• Location: Almost throughout renal tubul.

• Reabsorption retained drug back into the blood.

• Similar to absorption from the gut wall.

 Drug molecule passively transported through tubular reabsorption.

## Enhancing the excretion of drug in case of overdosing and poisoning

Drug existing in the non ionized lipid soluble state

completely reabsorbed

• Weakly acidic drug e.g salicylate excreted better if PH become alkaline and vice versa.

 Acidification or alkalization of urine with drug like ascorbic acid or sodium bicarbonate

used to hasten the excretion of drug in case of overdosing or poisoning.

## **Biliary excretion**

Minor rout of elimination for unmetabolized

![](_page_62_Figure_2.jpeg)

Major rout of elimination for metabolites eg.
 Water soluble conjugates like glucoronides.

![](_page_64_Figure_0.jpeg)

Prolong the action of drug

## **Pulmonary excretion**

• Important for gaseous and volatile liquid like general anasthetics.

Blood stream — Alveolar membrane
 Level Stream Alveolar membrane
 Level S

 Breathometer is often used to estimate the blood level of alcohol which correlate with degree of intoxication.

![](_page_66_Picture_1.jpeg)

![](_page_66_Picture_2.jpeg)

## Excretion in sewat, saliva, milk and gastric juice

Occur by passive diffusion of <u>non-ionized</u> form of drug.

Some are transported to suckling infant in breast milk

Drug used by **breast feeding** mother should be restricted to minimum.